



湖泊微生物硝化过程研究进展

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摘要: 湖泊中微生物介导的硝化作用在生境内部氮周转和温室气体 N_2O 释放方面扮演着关键的角色。因此, 研究湖泊微生物硝化过程及速率有助于我们整体评估湖泊生境内部的氮循环状态, 全面认识湖泊响应区域乃至全球气候变化的规律。本文综述了湖泊生境中硝化过程及其驱动微生物和影响因素, 包括氨氧化过程、亚硝酸盐氧化过程和完全氨氧化过程, 同时聚焦前沿, 归纳了氨氧化古菌、氨氧化细菌和完全氨氧化菌产生 N_2O 的机制和相对贡献。最后对湖泊硝化过程研究现状和未来发展方向提出总结和展望。

关键词: 湖泊, 微生物, 硝化, N_2O , 影响因素

硝化作用是指微生物将 NH_4^+ 氧化为 NO_3^- 的过程, 这一过程在很大程度上决定了湖泊无机氮库的形态分布, 并且影响着湖泊初级生产力, 在影响湖泊氮循环和控制温室气体 N_2O 释放方面扮演着关键角色^[1-3]。历史上, 我们对于硝化作用的认知一直在不断更新。20 世纪以来, 学术界普遍认为只有细菌(AOB)能够实现氨氧化; 直到 2005 年, 有学者在研究海洋^[4]和土壤^[5]的微生物基因组时发现古菌中存在类似的 *amoA* 基因, 并成功分离得到第一株氨氧化古菌(AOA) *Nitrosopumilus maritimus*^[6], 从而证明古菌也具有氨氧化功能; 于 2015 年, 完全氨氧化(comammox)

微生物被发现可以独立实现从氨到硝酸盐的氧化过程^[7-8], 极大地改变了我们对于硝化微生物的认识; 于 2019 年, AOA *Nitrosocosmicus oleophilus* 被发现在酸性条件下可表达一氧化氮还原酶 P450 NOR。经过计算, 当 pH 为 5.5 时, *N. oleophilus* 产生的 N_2O 中有超过 50% 来源于此生物途径^[9], 这一发现颠覆了 AOA 只能通过非生物途径产生 N_2O 的固有认识。在湖泊生态系统中, 硝化作用为反硝化作用源源不断地提供底物, 二者作为紧密联系的有机整体, 在降低湖泊氮负荷、减轻湖泊富营养化方面发挥着关键作用^[10-12]。因此, 研究硝化作用有助于我们全面认识湖泊氮循环的规

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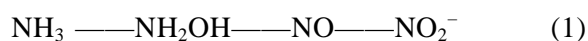
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律和生态意义。本文将综述湖泊生境中不同类型的硝化过程及相关的微生物群落组成, 并综合微生物代谢机理分析硝化过程的影响因素, 总结现有研究的不足, 展望未来研究的方向。

1 湖泊中的分步硝化过程

硝化作用可以通过分步实现, 主要包含氨氧化(反应式 1)和亚硝酸盐氧化(反应式 2)两个反应步骤。



1.1 氨氧化

氨氧化是分步硝化作用的第一步同时也是限速步骤。由于这一步骤依赖氨单加氧酶(AMO)来完成, 因此执行该步骤的微生物群落通常用编码 AMO 的 *amoA* 基因来鉴定。大多数 AOB 属于严格好氧的 β 变形菌纲和 γ 变形菌纲, 例如 *Nitrosomonas*、*Nitrospira* 和 *Nitrosococcus* 属^[13-15]。AOA 根据系统发育关系可分为以下属: *Nitrosopumilus*、*Nitrosotalea*、*Nitrosocaldus* 以及 *Nitrososphaera*^[16]。湖泊生境中的 AOB 主要为 *Nitrosospira*^[14]和 *Nitrosococcus*, 后者主要分布于盐湖中^[17-18]; AOA 主要为 *Nitrosotalea* 和 *Nitrosopumilus* 中的低盐度分支^[19]。在青藏高原盐湖中, Yang 等(2013)发现 AOA 主要为 *Nitrososphaera* 和 *Nitrosopumilus* 以及一个“低盐度分支”, 与世界其他湖泊 AOA 群落组成有着显著差异^[20], 暗示了盐湖生境中 AOA 分布具有地理差异的特殊性。AOA 与 AOB 具有不同的环境偏好和生态位, 受到湖泊生境中一系列环境因子的控制, 例如氨浓度、pH、氧浓度、盐度、硫化物浓度以及磷酸盐浓度等, 但本质上与二者生理

生化机制的差异有关(表 1)^[21-22]。AOA 具有极高的底物亲和力(半饱和常数 $K=133 \text{ nmol/L}$)和极低的底物临界值($\leq 10 \text{ nmol/L}$), 分别是 AOB 的 200 倍和 1/100, 二者数量级的差异使得 AOA 在底物极度匮乏的环境中占绝对优势^[21]。从分子机制上看, AOA 在进化过程中获得的 V-ATPase 十分关键, 该酶可以帮助 AOA 产生 ATP 和质子驱动力, 进而适应酸性、高盐、高压等逆境^[23-24]。具体到湖泊环境中, 大多数情况下, AOA 在种群丰度上均高于 AOB^[13,25-28]。但也存在相反的情形, 根据前人报道, 在青藏高原盐湖沉积物^[13]、极度富营养化的湖泊^[12]以及用于水产养殖的太湖东部湾^[25]等环境中, AOB 的丰度高于 AOA。目前鲜有学者建立起湖泊生境中 AOA、AOB 群落特征与 pH 的联系, 可能与大部分研究湖泊 pH 梯度较小且偏碱性的特点有关^[12,20,29]。在 pH 范围 4.8-7.7 的丹麦湖泊沉积物中 AOA 相对丰度被发现与 pH 呈负相关, 暗示了 AOA 对低 pH 的偏好性^[26]。纯培养研究显示, pH 降低会引起氨的质子化, 从而降低底物的生物可利用性, 使得耐受寡营养的 AOA 生存下来, 而 AOB 几乎不能在 pH 低于 6.5 的环境中进行氨氧化^[30-33]。另外, AOA 通常能够在高硫化物的条件下生存并表达氨氧化活性, 而 AOB 的生长则受到抑制, 这可能与 AOB 中含铜

表 1. AOA 与 AOB 的生理和生态分布差异

Table 1. Physiological and ecological distribution differences between AOA and AOB

Parameter	AOA	AOB
Substrate affinity	High	Low
Substrate threshold	Low	High
Ammonia	Low	High
pH	From acidic to alkaline	Near neutral pH
Oxygen	Low	High
Sulfide	Presence	Absence
Phosphate	Low	High

AMO 的合成受到抑制有关^[34-35]。湖泊溶解氧含量也是影响氨氧化速率和相关微生物的关键因素。纯菌动力学研究表明, AOA 对氧气的亲和力远远高于 AOB, 引起了二者对氧气的适应机制存在本质上的差异, 使得 AOA 在低氧环境中具有竞争优势^[21,36]。例如, 在盐湖环境中, 随着湖水深度增加至氧跃层, 溶解氧浓度大幅降低, 氨氧化速率的峰值也出现在溶解氧浓度较低的深层湖水和季节性低氧区, 并且与 AOA 丰度具有很好的相关性而与 AOB 丰度不相关, 暗示着 AOA 在低氧区对于氨氧化过程的实质性贡献^[37]。综上, AOA 往往在寡营养、低 pH、低氧和含硫的湖泊环境中比 AOB 更占优势。

盐度也是影响 AOA 与 AOB 丰度和多样性的一个重要因素, 但根据已有研究尚不能得出一致的规律。有研究显示, 高盐环境中 AOB 的丰度和活性较低而 AOA 多样性与氨氧化活力较高^[38-41]。因此, AOA 被认为在极端环境(如高盐湖泊)中更有竞争力, 而 AOB 的生长则受到限制^[21,42]。也有研究描述了不同的现象。如本课题组研究发现, 青藏高原盐湖沉积物中 AOB 在基因组和转录组层面的功能基因丰度均高于 AOA^[13]。从能量角度看, 高盐环境下微生物需要产生大量的能量来抵御盐度压力, 这种反应的热力学特征一定程度上决定了其生存的盐度上限^[43]。本课题组前期研究揭示了青藏高原不同盐度湖泊 AOA 的群落组成和丰度规律, 在盐度高达 160 g/L 的超盐湖中检测到了古菌 *amoA* 基因^[20], 却并未在这些湖泊水体中检测到可观的氨氧化速率(未发表数据)。考虑到青藏高原湖泊受到极强的太阳辐射, 影响着硝化微生物的生态位分布与活力^[44-45], 光抑制和盐度抑制共同作用是否导致氨氧化群落生

态位下移(比如在沉积物中表现出较高的活力)? 还需要更多深入的研究来揭示。

1.2 亚硝酸盐氧化

数十年来, 亚硝酸盐氧化过程受到的关注程度远不及氨氧化过程。造成这一现象的原因一方面可能是由于氨氧化被认为是硝化作用的限速步骤, 另一方面可能与亚硝酸盐氧化菌(NO₂-oxidizing bacteria, NOB)难以分离培养的特性有关^[46]。实际上, 亚硝酸盐的消耗途径直接决定了氮是以亚硝酸盐、硝酸盐以及铵盐的形式保存在湖泊生境中还是以一氧化氮、氧化亚氮或氮气形式逸散到大气中, 是影响湖泊氮收支的一个关键步骤。

根据系统发育 NOB 可以分为 *Nitrobacter*、*Nitrotoga*、*Nitrococcus*、*Nitrospira*、*Nitrospina*、*Nitrolancea* 与“*Candidatus Nitromaritima*” 7 个属, 属于 *Proteobacteria*、*Chloroflexi*、*Nitrospirae* 以及 *Nitrospinae* 4 个门^[47]。湖泊河流等淡水环境中的 NOB 主要为 *Nitrobacter*、*Nitrotoga*、*Nitrospira* lineage I 和 II; 盐湖等高盐环境中 NOB 则广泛分布于除 *Nitrolancea* 之外的 6 个属中。

亚硝酸盐氧化通过亚硝酸盐氧化还原酶(NXR)实现, 该酶通过将 2 个电子传递到呼吸链中来完成一次反应。NXR 与细胞质膜联系紧密, 由 NxrA、NxrB 与 NxrC 3 个亚基组成^[48-49]。负责与底物结合的 NxrA 亚基分为胞质型和胞浆型^[48], 前者分布于 *Nitrospira*^[48]、*Nitrospina*^[50] 和“*Candidatus Nitromaritima*”^[51], 后者分布于 *Nitrobacter*、*Nitrolancea* 和 *Nitrococcus* 中^[46,52]。胞质型的 NXR 能够在氧化亚硝酸盐的同时从水中获得氢离子, 从而获得质子动力势(proton motive force, PMF)并作为细胞的能量来源^[48]。具有胞浆型 NXR 的 NOB 在反应中产生质子却不能

贡献 PMF, 且需要消耗额外的能量用于硝酸盐与亚硝酸盐的膜内外传递^[47]。鉴于亚硝酸盐氧化产能较低, 这种微小的区别很可能成为这类 NOB 的生理瓶颈, 使其在特定的环境中处于劣势。NOB 具有丰富的代谢机制, 参与了湖泊中多种元素循环。例如: 在缺氧条件下, NOB 能够耦合甲酸盐氧化和硝酸盐还原并保持活性; 以氢气为能量来源, 还能够固定水体无机碳^[53-55]。值得注意的是, NOB 与氨氧化菌群常存在着共生和交叉饲养的现象: 首先, NOB 氧化环境中的尿素生成氨和 CO₂, 缺乏尿素降解酶的 AOA/AOB 从 NOB 获得氨, 并将其氧化为亚硝酸根, 随后 NOB 获得亚硝酸根并将其氧化, 实现双方生长的促进和硝化速率的提升^[54,56]。

NOB 群落结构与功能及其硝化速率取决于亚硝酸根浓度、无机碳浓度、温度以及盐度等环境因子。生理差异(胞浆型和胞质型 NXR, 以下简称为胞浆型 NOB 和胞质型 NOB)使不同 NOB 响应环境变化的机制和程度不同。NOB 对底物亲和力的差异^[47], 使不同类型 NOB 对底物浓度的响应不一: 底物亲和力高的胞质型 NOB(如 *Nitrospira lineage I*、*Nitrospina*)更加适应亚硝酸盐浓度低的环境, 如淡水湖泊; 而底物亲和力较低的胞浆型 NOB(如 *Nitrobacter*)则更加适应局部的或者季节性亚硝酸盐聚集的环境^[47,57-58]。无机碳的缺乏能够很大程度上抑制 *Nitrobacter* 的生长与活力; 但随着二氧化碳浓度的增加, *Nitrobacter* 的生长和活力又得到恢复, 这种现象可能与碳限制条件下该菌的碳浓缩机制有关^[59]。温度也是影响 NOB 不同种群分布的重要因素, *Nitrobacter*、*Nitrospira* 与 *Nitrotoga* 生长温度范围分别为 17-28 °C、10-28 °C 和 10-17 °C, 温度适应范围

的差异对解释不同 NOB 的季节性分布具有一定的指示作用^[60]。

在盐湖极端环境中, 氮循环微生物群落的生态位分布、组成和活力均受到盐度的影响, 尤其是产能较低的微生物^[43,61]。前人几乎没有在海水盐度以上的环境中观察到亚硝酸盐氧化的生物过程, 可能是由于 NOB 的生理生化特点使其难以在高盐环境中生存。在高盐环境中, 微生物需要大量的能量来抵御盐度压力, 而亚硝酸盐氧化过程产能(74 kJ/mol N)远小于硝化过程(275 kJ/mol N)和反硝化过程(800 kJ/mol N); 此外 NOB 需要大量能量来完成自养固碳, 约百倍于亚硝氧化过程的产能^[43,62-63]。然而, AOA(高达 160 g/L)和 γ -AOB(40-94 g/L)却能够在高盐环境中生存并保持氨氧化活性^[20,43,61]。理论上, 两步硝化反应热力学上的差异会引起某些高盐、超盐湖泊中亚硝酸根离子的积累。然而, 本课题组长期观测的青藏高原北部盐湖中并未检测到亚硝的积累^[13,20,64], 这暗示着亚硝通过其他途径被消耗, 如反硝化过程和厌氧氨氧化过程^[65]。如此一来, 盐湖中两步硝化过程的平衡可能被打破, 氨氧化可能与其他微生物建立了新的耦合机制, 值得深入探讨。

2 湖泊中的完全氨氧化过程

完全氨氧化作用(comammox)由 *Nitrospira lineage II* 实现。Comammox *Nitrospira* 曾经一直被认为是 NOB 的一员, 该微生物体内包含氨氧化(AMO、HAO)和亚硝酸盐氧化作用(NXR)的所有酶。Comammox 的氨单加氧酶在系统发育上与已知的 AMO 有着明显差异^[7], 据此可以将其分为 Clade A 和 Clade B 两支^[8]。其中 Clade A 可以进一步分为 Clade A.1 和 Clade A.2^[66]。根据目前

研究, 湖泊环境中的 Comammox 主要为 Clade A.1, 而 Clade A.2 和 Clade B 的丰度较低^[66-67]。

与分步硝化作用相比, Comammox 最大的优势体现在能量方面, 即利用同等的底物产能更多^[7], Comammox 反应途径决定了它生成 ATP 的效率更高^[68]。与 NOB *Nitrospira* 相比, Comammox *Nitrospira* 实现氨氧化与羟胺氧化的机制不同, 虽然无法同化亚硝酸盐、不能直接利用外部亚硝酸盐作为氮源, 却拥有更多样的尿素转运蛋白和内稳态基因, 因此能够独自实现从尿素降解到完全氨氧化的过程^[69]。与氨氧化菌相比, Comammox 的氨氧化和羟胺氧化基因与 β -AOB 最为相近, 暗示了它们在进化上可能存在关联^[7-8,69]。另外, Comammox 的基因组包含一种促进微生物低氧浓度下生长的基因, 使其在微氧条件下具有明显的竞争优势^[69-70]。与 AOA 类似, Comammox 缺少参与一氧化氮还原的基因, 不能完成大部分 AOB 具备的通过一氧化氮还原酶(NOR)产生 N_2O 的反应(表 2), 即硝化反硝化代谢过程。目前仅观察到 Comammox 通过非生物途径生成 N_2O , 生成量与 AOA 相当, 远低于 AOB^[71]。因此, 研究 Comammox 这一“绿色硝化菌”的分布规律及影响因素将对全球气候变化有一定的指示作用。

为了更好地了解这种新近发现的氮循环途径的生态学意义, 学者们对比传统硝化菌, 研究了 Comammox 分布规律、硝化过程中的相对贡献及其影响因素。基因组数据显示, Comammox 广泛分布于各种湖泊生境中(包括盐湖), 往往与 AOA 和 AOB 共存^[8,66-67,72-73]。目前在大范围湖泊生境中定量 Comammox 相对贡献的研究仍然欠缺。在武汉东湖与安徽巢湖沉积物样品中, Comammox 功能基因丰度高于 AOA 和 AOB^[72-73];

Xia 等(2018)综合宏基因组分析也得出了一致的结论, 即淡水生态系统中 Comammox 群落的相对丰度在所有氨氧化功能群中占据较高的比例(52.3%±37.5%), 暗示该过程对湖泊硝化作用不可忽视的贡献^[66]。影响 AOA 和 AOB 生态位分异的主要因素为氨亲和力的差异和 pH, 而 Comammox 需要与氨氧化菌竞争氨来维持生长。因此, 有学者推测氨浓度以及亲和力差异是影响它们分布的一个重要因素^[8,74]。动力学研究显示, Comammox *N. inopinata* 对氨的亲和力高于 AOB 和大部分 AOA, 能够在底物浓度极低(<5 $\mu\text{mol/L}$)的条件下竞争中胜出^[74-76]。此外, 氧气浓度也是十分重要的影响因素, 基于基因组的研究表明 Comammox 拥有特殊的低氧基因, 使其具有更高的氧亲和力; 前人在湖泊低氧环境中检测到 Comammox 丰度更高的现象, 暗示着 Comammox 比 AOA 和 AOB 更适应低氧气浓度的环境^[69-70,72]。值得注意的是, 数量上的优势并不一定等同于生化活力的高效性^[77]。因此, Comammox 在湖泊生境中的生态意义还需要深入探究。

3 湖泊硝化过程对温室气体 N_2O 的贡献

作为温室气体之一, N_2O 对全球变暖效应的贡献约占 6%。百年尺度上单分子 N_2O 的增温潜势(global warming potential)为 CO_2 的 298 倍^[78-79]。除温室效应外, N_2O 还能够在平流层进一步被氧化为 NO, 是导致臭氧层破坏的主要原因之一^[80]。 N_2O 作为一种反应副产物或中间产物, 全球 N_2O 排放量的 70% 来源于微生物参与的硝化和反硝化过程^[81-82]。

关于湖泊反硝化过程产 N_2O 机理和贡献的研究已经相对成熟, 而硝化过程则往往被忽视。实际上, AOA、AOB 和 Comammox 均能产生 N_2O 气体, 但其产生机制与产量却差别迥异。 N_2O 主要通过以下三种途径产生(表 2, 图 1): 1) 非生物途径, 也称杂交生成 N_2O 。在好氧条件下, AOA、AOB 和 Comammox 氨氧化的中间产物 NH_2OH 与 NO_2^- 化学反应生成 N_2O ^[83-84]。2) 硝化菌反硝化途径。在低氧条件下, AOA 和 AOB 通过亚硝酸盐还原酶 NirK 将 NO_2^- 还原为 NO, 实现反应的第一步; 实现第二步反应的酶有所差异: AOB 通过一氧化氮还原酶 NOR 产生 N_2O ; AOA 中没有编码这种酶的基因 *norB*, 而是在低 pH 下表达的 P450

NOR 催化下生成 N_2O ^[9,85-86]。3) AOB 独有的 CytL 蛋白催化途径。在厌氧条件下, CytL 氧化 NH_2OH 生成 N_2O ; 或在 NH_2OH 存在的条件下, 结合 NO 并将其还原为 N_2O ^[87]。尽管部分 Comammox 也含有类似 CytL 蛋白(相似度小于 55%), 却并没有发

表 2. 基于基因层面解读硝化细菌 N_2O 形成途径的差异

Table 2. Genes involved in N_2O -producing nitrifying pathways.

Gene	AOA	AOB	Comammox
<i>nirK</i>	+/-	+/-	+/-
<i>norCBQDSY</i>	-	+/-	-
<i>p450nor</i>	+/-	-	-
<i>cytL</i>	-	+/-	+/-

+/- denotes gene detected in several but not all organisms, - denotes not detected.

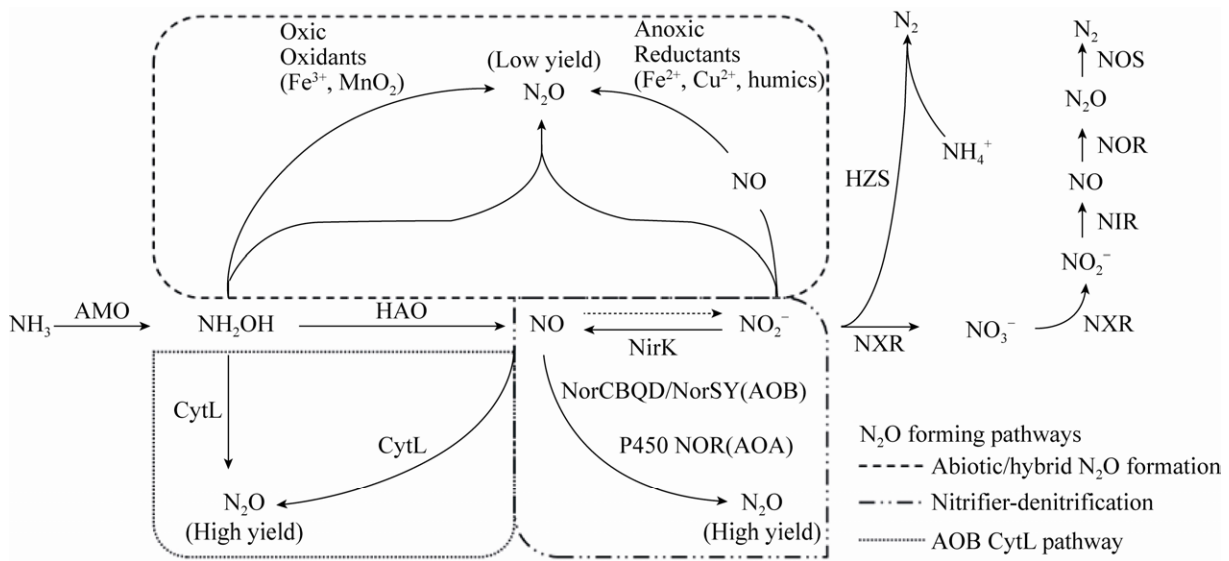


图 1. 氨氧化过程中 N_2O 形成的生物和非生物途径

Figure 1. Biotic and abiotic pathways involved in N_2O production derived from ammonia oxidation. Solid and dashed arrows depict confirmed and proposed reaction pathways, respectively. The enzymes or redox involved in the reaction are noted next to the arrow. AOB and Comammox contain AMO and HAO, while AOA does not contain HAO. The enzyme that catalyzes the oxidation of nitric oxide to nitrite is unknown. AMO: ammonia monooxygenase; HAO: hydroxylamine oxidoreductase; NirK: copper-containing nitrite reductase; NorCBQD/NorSY: nitric oxide reductase in AOB; P450 NOR: cytochrome P450 nitric oxide reductase in AOA; CytL: periplasmic tetraheme cyt. c P460 protein; NXR: nitrite oxidoreductase; NIR: nitrite reductase; NOR: nitric oxide reductase; NOS: nitrous oxide reductase; HZS: hydrazine synthase.

现 Comammox 通过该途径生成 N_2O ^[88]。纯培养和微宇宙实验一致显示, AOA 和 Comammox 贡献的 N_2O 产量低于 AOB, AOA 的 N_2O/NO_2^- 比值为 0.04%–0.07%, AOB 则为 0.095%–0.270%^[84,89–91]。综合来看, 与生物成因相比, 非生物成因的 N_2O 占比极低, 不到 3%; 且受 pH 影响很大, pH 越低, 化学生成 N_2O 的速率越高^[92]。综合来看, 硝化过程产生的 N_2O 主要来自于 AOB, 但目前的研究结果集中在纯培养实验层面, 在实际湖泊生境中硝化过程对温室气体 N_2O 的贡献如何, 如何定量区分硝化过程和反硝化过程的 N_2O 生成比例是当前湖泊硝化过程研究的前沿问题。Zhang 等(2015)提出了 N_2O 排放的三通道模型, 并通过同位素成对标记和来源区分法解析了 N_2O 的来源, 结果表明除了反硝化过程, 自养硝化和异养硝化过程也是重要的 N_2O 产生途径, 甚至在某些陆地生态系统中超过反硝化作用, 主导了 N_2O 的产生^[93–94]。将该技术应用于湖泊生境有助于全面认识硝化过程对温室气体 N_2O 的贡献, 深刻理解硝化过程在全球气候变化中扮演的角色。

4 总结和展望

尽管湖泊中硝化菌群的生理生态特征各不相同, 既分工合作又相互竞争, 但它们最终作为一个整体完成了湖泊的硝化过程。湖泊生境作为氮循环的热点区域, 氮转化速率整体上高于河口和

海洋等水生系统(表 3)^[1,95]。总体来说, 环境条件控制着硝化过程发生的位置和速率。硝化过程很大程度上依赖于氧气和氨的生物可利用性, 还受到 pH、硫化物浓度、温度、盐度以及光等环境因子的限制^[95]。一般来说, 底物充足的富营养湖泊比寡营养湖泊的硝化速率要高, 这与 AOB 的丰度和活力密切相关^[12,26]。湖泊水体硝化速率的峰值往往出现在深层, 受到氧气浓度、光抑制作用和浮游生物竞争机制的综合影响^[1,37,96]。水体酸化会显著降低硝化速率, 在 pH 小于 3 的湖泊中硝化作用被完全抑制^[97–98]; 大多数化能无机自养型硝化菌在中性偏微碱性环境中(pH 7.0–8.5)的生长和硝化活力最佳^[99]; 极端的碱性条件并不会完全抑制沉积物中的硝化作用, 如 *Nitrobacter alkalicus* 能够在 pH 高达 10 的苏打湖沉积物中生存^[100], 这可能与沉积物中粘土矿物的缓冲作用有关^[101–102], 而这种机制也一定程度上解释了沉积物(10^5 – 10^7 cells/cm³)比湖泊水体(10^1 – 10^4 cells/cm³)的硝化菌群和硝化速率高出几个数量级的现象(表 3)^[95,99,103]。

综合湖泊硝化过程与其驱动微生物群落的研究发现, 微生物参与的硝化过程很大程度上决定了湖泊无机氮库的赋存形式和分布规律, 在湖泊内部氮循环和温室气体 N_2O 的释放方面起着不可忽视的作用, 对区域乃至全球气候变化有着深远的影响。AOA、AOB、Comammox 等硝化微生物

表 3. 湖泊与其他水生系统硝化速率范围^[95]

Table 3. Nitrification rate ranges in lakes and other aquatic ecosystems^[95]

Ecosystem	Lake		River		Coastal marine		Deep ocean	
	Sed	Water	Sed	Water	Sed	Water	Sed	Water
Rates	2–20	0–1	3–23	0–4	0.2–7.0	0.002–0.200	0.003–0.100	0.001–0.010

Sed denotes Sediment, coastal marine and deep ocean are divided according to maximum depth of less than 100 m and above 100 m. Units for sediment nitrification rates are mmol N/(m²·d), for water nitrification rates are mmol N/(L·d).

新代谢途径的发现大大拓宽了传统观念中硝化作用的范围,加深了我们对于硝化作用响应气候变化的理解。但通过总结前人的研究和本课题组相关数据可以发现,仍有一些值得深入探究的科学问题,例如:(1)本课题组前期研究观察到青藏高原盐湖中氨氧化速率和氨氧化微生物群落分布规律不一致的现象,考虑到青藏高原湖泊受到极强的太阳辐射,光和盐度共同抑制作用将如何影响着硝化微生物的生态位分布与活力^[44-45],值得进一步探究;(2)亚硝酸盐氧化菌(0-35 g/L)和氨氧化菌(0-160 g/L)在盐度适应范围方面存在差异^[43,61-62],这种差异是否会引起盐湖中两步硝化过程的失衡?氨氧化微生物是否与其他微生物建立了新的耦合机制(如厌氧氨氧化、反硝化与氨氧化过程的耦合)?(3)定量区分硝化过程和反硝化过程的 N_2O 生成比例是当前湖泊硝化过程研究的前沿问题,应用同位素成对标记技术和来源区分计算方法将有助于全面认识硝化过程在全球气候变化中的贡献。这些问题目前尚无定论,但解决上述问题无疑可以帮助我们更好地理解各个环境中的氮循环过程及其环境效应。

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Research progress in microbial nitrification in lakes

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Abstract: Microbially mediated nitrification plays a key role in nitrogen transformation and emission of greenhouse gases N_2O in lacustrine ecosystems. Thus, studying microbial nitrification process and rate in lake environments helps comprehensively understand nitrogen biogeochemical cycle in lacustrine ecosystems and its response to regional and global climate change. This review summarized processes and microbial community compositions of nitrification (including ammonia oxidation, nitrite oxidation and complete ammonia oxidation) in lacustrine ecosystems and their influencing factors. Spotlight was given to the mechanisms and relative contributions of N_2O production through ammonia oxidizing archaea and bacteria and complete ammonia oxidizing bacteria, followed by summary and prospect on the current situation and future development of nitrification researches in lakes.

Keywords: lakes, microbes, nitrification, N_2O , influencing factors

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